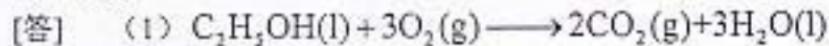


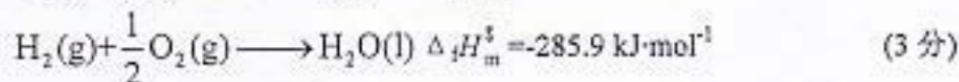
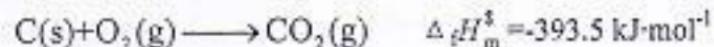
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20. 10 分



$$\Delta_c H_m^\ddagger = -1366.9 \text{ kJ}\cdot\text{mol}^{-1}$$



$$(2) \Delta_c H_m^\ddagger = -1366.9 \text{ kJ}\cdot\text{mol}^{-1} = 2 \Delta_f H_m^\ddagger (\text{CO}_2) + 3 \Delta_f H_m^\ddagger (\text{H}_2\text{O}) - \Delta_f H_m^\ddagger (\text{C}_2\text{H}_5\text{OH})$$

$$\text{得 } \Delta_f H_m^\ddagger (\text{C}_2\text{H}_5\text{OH}) = -277.8 \text{ kJ}\cdot\text{mol}^{-1} \quad (3 \text{ 分})$$

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(3)  $Q_V = Q_p - \Delta n RT = 2 \Delta_c H_m^\ddagger + 2RT = -2728.8 \text{ kJ}\cdot\text{mol}^{-1} \quad (4 \text{ 分})$